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Key indicators

Single-crystal X-ray study

T = 294 K

Mean $\sigma(\text{C}-\text{C}) = 0.006 \text{ \AA}$

R factor = 0.084

wR factor = 0.180

Data-to-parameter ratio = 14.1

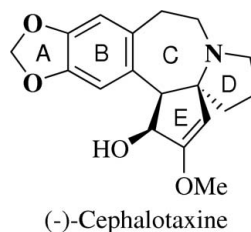
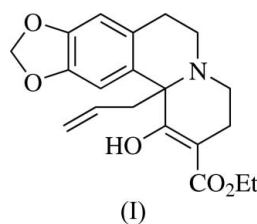
For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Ethyl 12b-allyl-1-hydroxy-3,6,7,12b-tetrahydro-
4H-1,3-dioxolo[4,5-g]pyrido[2,1-a]isoquinoline-
2-carboxylateThe title compound, $\text{C}_{20}\text{H}_{23}\text{NO}_5$, is an advanced intermediate
used in the total synthesis of cephalotaxine. An intramolecular
 $\text{O}-\text{H}\cdots\text{O}$ hydrogen bond exists in the crystal structure.

Received 26 July 2006

Accepted 13 August 2006

Comment

Cephalotaxine (CET), parent of the cephalotaxus alkaloids (Huang & Xue, 1984; Jalil Miah *et al.*, 1998), possesses a unique benzazepine-bearing pentacyclic *ABCDE* ring skeleton. Naturally occurring esters of CET (harringtonine and homoharringtonine) have been found to be highly effective in the treatment of acute human leukaemia and are currently undergoing advanced clinical trials (Kantarjian *et al.*, 2001). Homoharringtonine is also a potent agent against strains of the chloroquine-resistant *Plasmodium f. malaria* parasite *in vitro* (Whaun & Brown, 1990). The unique structure and therapeutic potential of this group of alkaloids have stimulated much synthetic research, which has produced several elegant total syntheses of CET (Tietze & Schirok, 1999; Koseki *et al.*, 2002; Suga *et al.*, 2002), and numerous studies of the construction of the pentacyclic ring system.



The title enol ester, (I), an advanced intermediate used in the total synthesis of CET, was obtained from a Dieckmann cyclization of the corresponding diester precursor (Wang, 2003). As shown in (Fig. 1), the molecule has reactive side groups that could serve as handles for further conversion to cephalotaxine (CET). The two six-membered heterocyclic rings are *trans*-fused. The six-membered heterocycle bearing the CO_2Et group adopts a twist conformation, while the other adopts a half-chair conformation. In the molecule, atom O5 acts as an intramolecular hydrogen-bond donor, forming an

O—H...O bond with the carbonyl atom O3. The displacement parameters for C16 and C17 are quite large, which could explain the too short C16—C17 distance (1.093 Å).

Experimental

To a refluxing mixture of KO^tBu (134 mg, 1.2 mmol) in 10 ml of toluene was added the corresponding allyl diester of the title compound (403 mg, 1.0 mmol) in 2 ml of toluene. After 30 min, the reaction mixture was diluted with 20 ml of EtOAc, washed with water (5 ml) and brine (5 ml), dried, and evaporated under reduced pressure. The residue was purified by silica-gel column chromatography, eluted with petroleum ether–EtOAc (5:1 v/v), to give the corresponding enol ester, (I) (268 mg, 75%), as colourless needles. Single crystals suitable for X-ray determination were obtained by slow evaporation of an AcOEt solution over a period of several days (m.p. 395–396 K). HRMS (EI) *m/z* calculated for [M + H] C₂₀H₂₄NO₅: 358.1649; found: 358.1653.

Crystal data

C ₂₀ H ₂₃ NO ₅	Z = 4
<i>M_r</i> = 357.39	<i>D_x</i> = 1.321 Mg m ⁻³
Monoclinic, <i>P</i> ₂ ₁ / <i>c</i>	Mo <i>K</i> α radiation
<i>a</i> = 8.5365 (7) Å	<i>μ</i> = 0.10 mm ⁻¹
<i>b</i> = 20.0523 (18) Å	<i>T</i> = 294 (2) K
<i>c</i> = 10.6574 (9) Å	Plate, colourless
<i>β</i> = 99.854 (4)°	0.29 × 0.13 × 0.06 mm
<i>V</i> = 1797.4 (3) Å ³	

Data collection

Bruker SMART CCD area-detector diffractometer	3350 independent reflections
<i>φ</i> and <i>ω</i> scans	1915 reflections with <i>I</i> > 2σ(<i>I</i>)
Absorption correction: none	<i>R</i> _{int} = 0.054
9495 measured reflections	<i>θ</i> _{max} = 25.5°

Refinement

Refinement on <i>F</i> ²	$w = 1/[\sigma^2(F_o^2) + (0.0336P)^2 + 2.3489P]$
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)] = 0.084	where $P = (F_o^2 + 2F_c^2)/3$
<i>wR</i> (<i>F</i> ²) = 0.180	(<i>Δ</i> / <i>σ</i>) _{max} < 0.001
<i>S</i> = 1.16	<i>Δρ</i> _{max} = 0.60 e Å ⁻³
3350 reflections	<i>Δρ</i> _{min} = -0.40 e Å ⁻³
238 parameters	Extinction correction: <i>SHELXL97</i>
H-atom parameters constrained	Extinction coefficient: 0.0022 (9)

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O5—H5...O3	0.82	1.83	2.556 (4)	147

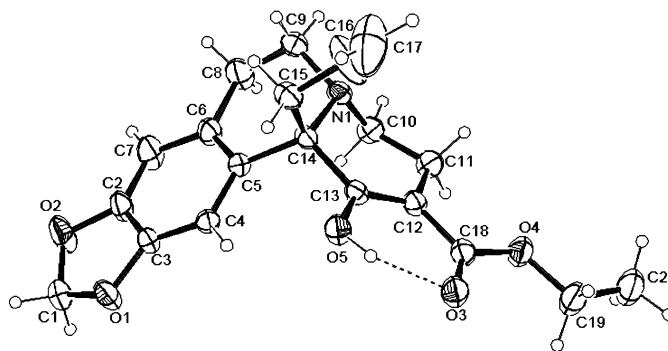


Figure 1

The structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. The dashed line indicates the intramolecular hydrogen bond.

All H atoms were positioned geometrically (C—H values were set at 0.97, 0.96, 0.93 and 0.82 Å for CH₂, CH₃, CH and OH H atoms, respectively) and refined with a riding model, with *U*_{iso}(H) = 1.2 or 1.5 times *U*_{eq}(C), or 1.5*U*_{eq}(O). No attempt was made to model a disorder for the CH₂CH=CH₂ group.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000); software used to prepare material for publication: *SHELXTL*.

We acknowledge financial support from the Research Fund for the new faculty at the State Key Laboratory of Applied Organic Chemistry.

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